

TOTAL SYNTHESIS OF (\pm)-ISOPETASOL, (\pm)-3-EPIISOPETASOL,
AND (\pm)-WARBURGIADION

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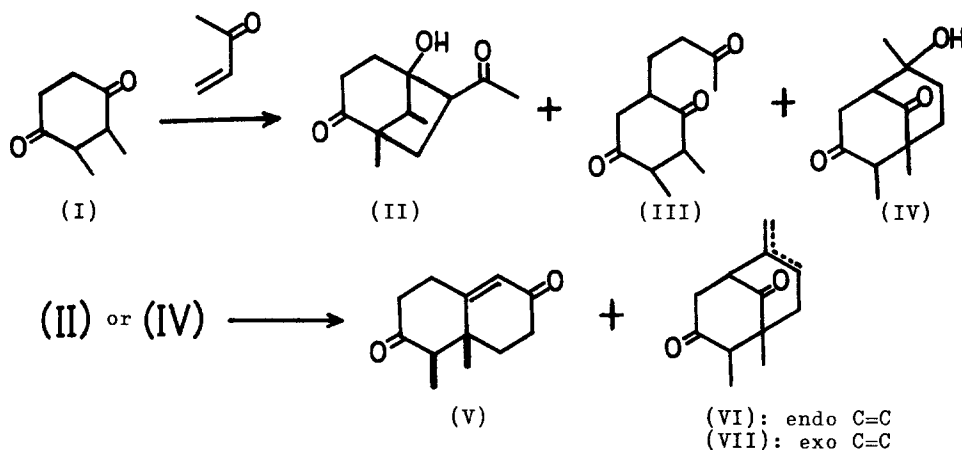
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The eremophilane type sesquiterpenes are characterized by a carbon framework which does not obey the well known isoprene rule. There are interesting aspects of the eremophilanes from a synthetic standpoint, namely the vicinal cis-dimethyl substituents on the hydronaphthalene framework.

We will report the total synthesis of (\pm)-isopetasol, isolated from Petasites officinalis^{1,2)} and P. japonicus Maxim,³⁾ (compositae), and (\pm)-3-epiisopetasol, and (\pm)-warburgiadion, isolated from Warburgia ugandensis Sprague (canellaceae).

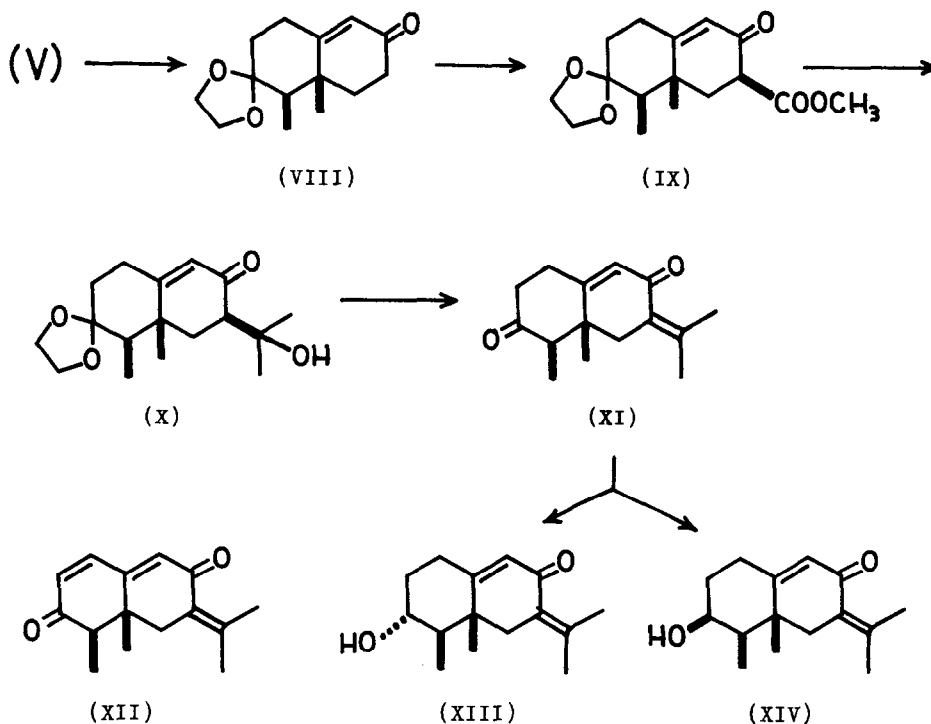
The bicyclic key intermediate (V) was prepared by the Robinson annelation reaction of 2,3-dimethylcyclohexane-1,4-dione (I) with methyl vinyl ketone in the presence of sodium ethoxide at -20° for 98 hr. The reaction products were chromatographed to give three intermediates; 7-acetyl-5,8-dimethyl-4-oxo-bicyclo[3.2.1]octane-1-ol (II), m.p. 145-146° [i.r. cm^{-1} : 3400 (OH), 1705 and 1690 (C=O); n.m.r. δ : 0.82 (d, J=7.5 Hz; 3H), 1.03 (s; 3H), 2.24 (s; 3H), and 3.35 (s; 1H)], 2,3-dimethyl-6-(3'-oxobutyl)cyclohexane-1,4-dione (III), m.p. 92-94° [i.r. cm^{-1} : 1710 (C=O); n.m.r. δ : 1.15 (d, J=7.0 Hz; 6H), 2.12 (s; 3H)], and 4,5,8-trimethyl-3,9-dioxo-bicyclo[3.3.1]nonane-8-ol (IV), m.p. 163-165° [i.r. cm^{-1} : 3420 (OH), 1710 (C=O); n.m.r. δ : 1.11 (s; 3H), 1.14 (d, J=7.5 Hz; 3H), 1.37 (s; 3H), and 2.12 (s; 1H)]. Treatment of (II) and (IV) with p-toluene sulfonic acid in refluxing benzene for 20 hr gave the enone (V), m.p. 101-103° (63% yield) [i.r. cm^{-1} : 1710 and 1680 (C=O), 1630 (C=C); n.m.r. δ : 1.04 (d, J=7.5 Hz; $\text{C}_4\text{-CH}_3$), 1.07 (s; $\text{C}_5\text{-CH}_3$), and 5.87 (s; $\text{C}_9\text{-H}$)] which may possess cis-dimethyl, together with a mixture of olefines (VI) and (VII) (10-15% yield), respectively.



Enone (V) and a mole of ethylene glycol in benzene containing a catalytic amounts of *p*-toluene sulfonic acid were refluxed for 20 hr and a mono-ketal (VIII), m.p. 80–83° was obtained. Reaction of (VIII) with dimethylcarbonate in the presence of sodium hydride in abs. dioxane gave an enone ester (IX), m.p. 158–160°, in quantitative yield. The methoxycarbonyl substituent possess equatorial (7 β) orientation, which showed at δ 3.50 (C₇-H; dd, $J=6.3$ and 11.3 Hz) in their n.m.r. spectrum. Treatment of (IX) with an excess of ethereal methyl lithium in the presence of sodium hydride afforded enone alcohol (X), m.p. 91–92°. Dehydration and deketalisation of (X) by treatment with 1% hydrochloric acid in refluxing methanol produced a keto-isopropylidene derivative (XI), m.p. 110–112° [i.r. cm^{-1} : 1720 and 1680 (C=O), 1640 and 1630 (C=C); n.m.r. δ : 0.96 (s; C₅-CH₃), 1.08 (d, $J=7.5$ Hz; C₄-CH₃), 1.85 and 2.13 (s; $\begin{array}{c} \text{CH} \\ \diagdown \quad \diagup \\ \text{CH}_2 \end{array}$), and 5.92 (s; C₉-H)]. The n.m.r. data of our synthetic (XI) was in agreement with that for natural isopetasone as reported by Brooks and Draffan⁴⁾ and Kurihara *et al.*⁵⁾ Brooks and Draffan⁴⁾ have reported the conversion of natural isopetasone into (+)-warburgiadion (XII), isolated from the heartwood of *Warburgia ugandensis* Sprague (Canellaceae), by dehydrogenation with 2,3-dichloro-5,6-dicyanobenzoquinone in dioxane. Therefore, our synthesis of (XI) constitutes a formal total synthesis of (\pm)-warburgiadion.

Reduction of (XI) with sodium borohydride gave (\pm)-isopetasol (XIII), m.p. 105–106° [i.r. cm^{-1} : 3450 (OH), 1670 (C=O), and 1635 (C=C); u.v. nm: $\lambda_{\text{max}}^{\text{EtOH}}$ 247

(ϵ 12,000) and 279 (ϵ 7,200); n.m.r. δ : 0.98 (s; C₅-CH₃), 1.10 (d, J=7.5 Hz; C₄-CH₃), 1.85 and 2.10 (s; $\begin{smallmatrix} \text{CH} \\ \text{CH}_3 \end{smallmatrix}$), 3.58 (m, W1/2=25 Hz; C₃-H), and 5.78 (s; C₉-H)] and (\pm)-3-epiisopetasol (XIV), m.p. 109-110° [i.r. cm⁻¹: 3450 (OH), 1660 (C=O), 1630 and 1610 (C=C); u.v. nm: $\lambda_{\text{max}}^{\text{EtOH}}$ 248 (ϵ 11,200) and 279 (ϵ 7,000); n.m.r. δ : 1.13 (d, J=7.5 Hz; C₄-CH₃), 1.18 (s; C₅-CH₃), 1.85 and 2.10 (s; $\begin{smallmatrix} \text{CH} \\ \text{CH}_3 \end{smallmatrix}$), 3.95 (m; C₃-H), and 5.80 (s; C₉-H)] in the ratio of 1:5. I.r., u.v., n.m.r. spectra and gas and thin-layer chromatographic data confirmed the identity of the synthetic compound (XIII) with natural (+)-isopetasol which was isolated from *Petasites japonicus* Maxim by Naya and his coworkers.³⁾ The stereoformula of the alcohol (XIV) was also confirmed to be (\pm)-3-epiisopetasol by its physico-chemical data.



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References

- 1) A. Aebi, J. Büchi, T. Waarler, E. Eichenberger, and J. Schmutz, Pharm. Acta Helv., 30, 277 (1955).
- 2) A. Aebi and C. Djerassi, Helv. Chim. Acta, 42, 1785 (1959);
D. Herbest and C. Djerassi, J. Amer. Chem. Soc., 82, 4337 (1960).
- 3) K. Naya, I. Takagi, Y. Kawaguchi, and Y. Asada, Tetrahedron, 24, 5871 (1968).
- 4) C.J.W. Brooks and G.H. Draffan, Chem. Commun., 701 (1966).
- 5) T. Kurihara, K. Ro, H. Takada, and H. Ito, Tohoku Yakka Daigaku Nenpo, 13, 75 (1960).